

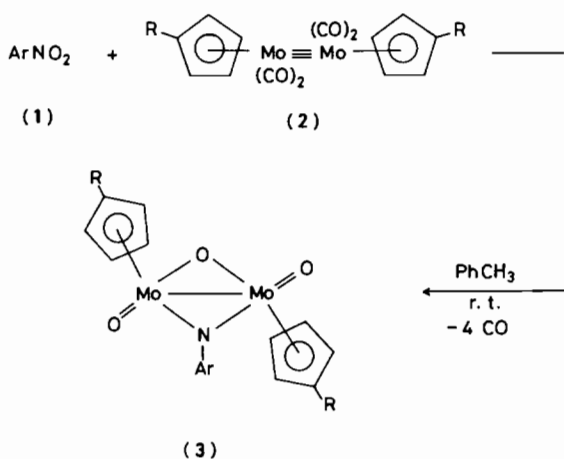
Reactions of Nitrosoarenes with the Molybdenum–Molybdenum Triple Bond

JEAN-FRANÇOIS PETRIGNANI and HOWARD ALPER*

Ottawa-Carleton Institute for Research and Graduate Studies in Chemistry, Department of Chemistry, University of Ottawa, Ont. K1N 9B4, Canada

Received July 5, 1983

Recently we described the novel reaction of nitrosoarenes (1) with cyclopentadienylmolybdenum dicarbonyl dimer (2, R = H) and related complexes containing a metal–metal triple bond. The products of these reactions are complexes (3) having bridging and

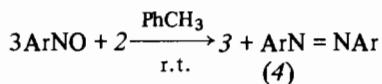


*Author to whom correspondence should be addressed.

terminal oxygen ligands. These complexes result from complete decarbonylation of 2 [1].

It seemed conceivable to us that the reaction may proceed *via* initial deoxygenation of the nitro compound to a nitrosoarene. If correct, then nitrosobenzenes should react with 2 to afford the same class of complexes. We now report the realization of this process.

Treatment of nitrosobenzene and *o*-nitrosotoluene with 2, R = H in toluene at room temperature afforded 3, R = H. The same type of complex was obtained using the methylcyclopentadienyl complex 2, R = CH_3 , as the reactant. Azobenzenes (4) were by-products of these reactions. The complexes (3) obtained from nitrosobenzenes gave the same characteristic infrared and nuclear magnetic resonance spectra (Table I) as those previously prepared from nitroarenes [1].



In conclusion, nitrosobenzenes may be intermediates in the conversion of nitro compounds to the binuclear molybdenum oxo complexes (3).

Acknowledgement

We are grateful to the Natural Sciences and Engineering Research Council for support of this work.

References

- 1 H. Alper, J. F. Petrigani, F. W. B. Einstein and A. C. Willis, *J. Am. Chem. Soc.*, 105, 1701 (1983).

TABLE I. Spectral Properties of 3^a.

2, R =	ArNO	IR ^b		NMR ^c δ ppm
		$\nu_{\text{Mo}=\text{O}}$ cm^{-1}	$\nu_{\text{Mo}-\text{O}}$	
H	PhNO	898	815	6.03(s, 10H, C ₅ H ₅) 7.10–7.90(m, 5H, Ph)
H	<i>o</i> -CH ₃ C ₆ H ₄ NO	904	811	2.53(s, 3H, CH ₃), 5.86(s, 5H, C ₅ H ₅), 5.96(s, 5H, C ₅ H ₅), 7.20(m, 4H, benzenoid protons)
CH ₃	PhNO	896	809	1.90(s, 6H, CH ₃), 5.43, 5.73, 6.00(m, 8H, cyclopentadienyl ring protons), 7.05–7.90(m, 5H, benzenoid protons)
CH ₃	<i>o</i> -CH ₃ C ₆ H ₄ NO	895	812	1.98(s, 6H, CH ₃), 2.53(s, 3H, CH ₃), 5.43, 5.80, 6.03(m, 8H, cyclopentadienyl ring protons), 7.25(m, 4H, benzenoid protons)

^aSatisfactory C, H, N analyses were obtained for 3. ^bKBr disc. ^cCDCl₃ with tetramethylsilane as internal standard.